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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR ATTORNEY DOCKET		CONFIRMATION NO.
10/813,529	03/30/2004	Tadahiko Kubota	09792909-5847	6143
	7590 05/15/200 EIN NATH & ROSEN'	EXAMINER		
P.O. BOX 0610		ECHELMEYER, ALIX ELIZABETH		
WACKER DRIVE STATION, SEARS TOWER CHICAGO, IL 60606-1080			ART UNIT	PAPER NUMBER
			1795	
			MAIL DATE	DELIVERY MODE
			05/15/2009	PAPER

# Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

		Application N	<b>D</b> .	Applicant(s)		
Office Action Summary		10/813,529		KUBOTA ET AL.		
		Examiner		Art Unit		
		Alix Elizabeth B	Echelmeyer	1795		
The MAILING DATE of the Period for Reply	is communication ap	pears on the cov	er sheet with the c	orrespondence ac	ddress	
A SHORTENED STATUTORY WHICHEVER IS LONGER, FR - Extensions of time may be available unde after SIX (6) MONTHS from the mailing d If NO period for reply is specified above, t - Failure to reply within the set or extended Any reply received by the Office later thar earned patent term adjustment. See 37 C	OM THE MAILING D r the provisions of 37 CFR 1.1 ate of this communication. the maximum statutory period period for reply will, by statut- three months after the mailing	DATE OF THIS ( 136(a). In no event, ho will apply and will expite, cause the application	COMMUNICATION wever, may a reply be time of SIX (6) MONTHS from to become ABANDONE	<b>1.</b> hely filed the mailing date of this c ○ (35 U.S.C. § 133).		
Status						
Responsive to communication is FINAL.  3) Since this application is in closed in accordance with	2b)∏ This n condition for allowa	s action is non-f	ormal matters, pro		e merits is	
Disposition of Claims						
4)	is/are withdra owed. re rejected. ected to.	awn from conside				
Application Papers						
9) The specification is object  10) The drawing(s) filed on  Applicant may not request the Replacement drawing sheet  11) The oath or declaration is	is/are: a) ☐ acc nat any objection to the (s) including the correc	cepted or b) cepte	ld in abeyance. See the drawing(s) is obj	e 37 CFR 1.85(a). ected to. See 37 C	, ,	
Priority under 35 U.S.C. § 119						
<ul> <li>12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).</li> <li>a) All b) Some * c) None of:</li> <li>1. Certified copies of the priority documents have been received.</li> <li>2. Certified copies of the priority documents have been received in Application No</li> <li>3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).</li> <li>* See the attached detailed Office action for a list of the certified copies not received.</li> </ul>						
Attachment(s)  1) Notice of References Cited (PTO-8922) Notice of Draftsperson's Patent Draw 3) Information Disclosure Statement(s) (Paper No(s)/Mail Date	ing Review (PTO-948)	4) [ 5) [ 6) [	Interview Summary Paper No(s)/Mail Da Notice of Informal P Other:	ite		

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#### **DETAILED ACTION**

### Response to Amendment

This Office Action is in response to the amendment filed February 12, 2009.
 Claim 1 has been amended. Claims 2, 4 and 9-18 were previously cancelled. Claims 1,
 3 and 5-8 are pending and are rejected finally for the reasons given below.

### Claim Rejections - 35 USC § 103

- 2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 3. Claims 1, 3 and 5-8 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sonoda et al. (US 2002/0028389) in view of Okamoto et al. (US 2003/0027050).

Sonoda et al. teach a non-aqueous electrolyte for use in an electrochemical device such as a lithium battery (abstract, [0001]). As for claim 1, it is well known to those having ordinary skill in the art that a lithium battery contains a cathode, an anode and an electrolyte.

The electrolyte of Sonoda et al. contains a solute represented by formula (1):  $MBR^1R^2R^3R^4$  (abstract, [0010]).  $R^1$ ,  $R^2$ ,  $R^3$  and  $R^4$  may be represented by  $C_nF_{2n+1}$  or  $C_mF_{2m+1}SO_2$  ([0011]). Additionally, since formula (1) is in solution in the electrolyte, it can be considered as its cation and anion:  $M^+$  ( $BR^1R^2R^3R^4$ ) $^-$  ([0019]).

A specific example of the material represented by formula (1) includes LiB(CF<sub>3</sub>)<sub>4</sub> ([0012]), which is identical to the material disclosed in the instant specification (see paragraph 5, above).

As for the limitation concerning the cathode active material, Sonoda et al. disclose that the positive active material is a transition metal complex oxide ([0051]).

As for claim 3, examples of the negative material include carbon materials, TiS<sub>2</sub>, and alkali metals such as silicon ([0044]).

Regarding claims 6-8, the electrolyte of Sonoda et al. may also include additional anions such as one or a mixture of  $PF_6^-$ ,  $BF_4^-$ ,  $ClO_4^-$ ,  $AsF_6^-$  or  $N(CF_3SO_2)_2^-$  ([0068]).

As for claim 5, Sonoda et al. fail to teach that the moisture content in the electrolyte is 100 ppm or less at a mass ratio in relation to the electrolyte.

Sonoda et al. teach that too much moisture in the electrolyte causes it to decompose ([0004]).

It would have been obvious to one having ordinary skill in the art at the time the invention was made to find the lowest tolerable amount of moisture in the electrolyte to prevent decomposition, since it has been held that discovering an optimum value of a result effective variable involves only routine skill in the art. MPEP 2144.05 (II B).

With further regard to claim 1, Sonoda et al. teach silicon as the anode active material ([0044]) but fail to teach that the anode active material is a silicon thin film.

Okamoto et al. teach the use of a silicon thin film as the anode active material in a rechargeable lithium battery ([0037]-[0038]). Okamoto et al. teach that silicon is preferable to other materials, and that it is capable of storing lithium via alloying.

Okamoto et al. teach that the thin film may be made by CVD or sputtering ([0039]).

According to Applicant's arguments, filed February 12, 2009, the gas phase deposition method inherently inhibits destruction by expansion or shrinkage of the anode material and forms an alloy between at least part of the interface between the active material layer and the anode current collector (see page 6, first full paragraph).

It would be desirable to make a silicon thin film anode active material layer in the battery of Sonoda et al. such as taught by Okamoto et al. since the skilled artisan would recognize that the anode active material could be made with a thin film, reducing the weight, size, and energy density of the battery.

Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to make a silicon thin film anode active material layer in the battery of Sonoda et al. such as taught by Okamoto et al. since the skilled artisan would recognize that the anode active material could be made with a thin film, reducing the weight, size, and energy density of the battery.

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## Response to Arguments

4. Applicant's arguments with respect to claims 1, 3 and 5-8 have been considered but are most in view of the new ground of rejection, see above. The new grounds of rejection were necessitated by amendment.

#### Conclusion

5. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Alix Elizabeth Echelmeyer whose telephone number is (571)272-1101. The examiner can normally be reached on Mon-Fri 8-5:30.

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan can be reached on 571-272-1292. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/PATRICK RYAN/
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